# Cobalt-Free Cathodes for Next Generation Lithium-Ion Batteries Project ID: bat417

# Principal Investigator: Neil J. Kidner



## **OVERVIEW**

#### **Budget**

Total Project Funding: \$3.08 M

- ▶ DOE share: \$2.46 M
- Cost share: \$620 k
- Funding for FY2020: \$1.1 M
- Funding for FY2021: \$1.0 M
- Funding for FY2022: \$1.0 M

#### <u>Timeline</u>

- Project Start Date: Jan. 2019
- Project End Date: Mar. 2022
- Percent Complete: 100 %

#### Partners

- Ohio State University: Dr. Jung-Hyun Kim Battery testing Cell chemistry development
- Navitas Systems Dr. James Dong Large-scale electrode fabrication

#### Barrier and Technical Targets

Cycle Life: 1000 cycles C/3 deep discharge with < 20 % energy fade

2-Ahr battery manufacture and testing

Cost: < \$100/kWh</p>

### RELEVANCE

- ▶ The increased demand for EVs is driving demand for battery materials.
- ▶ Renewed interest in reduced/cobalt free Li-ion battery cathode formulations
- ▶ Opportunity to reestablish U.S. dominance in batteries

#### <u>Objective</u>

**Impact** 

- ▶ Develop high performance and cobalt-free Li-ion battery based on high voltage lithium manganese nickel titanium oxide (LNMTO) cathode and complementary cell chemistry (electrolyte/cathode formulation)
- ▶ Identify low-cost, scalable process for producing cathode powder
- ▶ Identify strategy for Nexceris to grow domestic manufacturing and create jobs to support new clean energy and e-mobility opportunities

#### **APPROACH** TiOx-enriched inorganic SEI Core-shell microstructure TiOx enriched surface shell Self healing Low Ti substitution in core In-situ formation Passivation of active material Electrolyte LiPAA polymeric SEI In-situ formation LNMTO Passivation of active materia and carbon Extra Li+ donation Proton scavenge

Develop cobalt-free cell based on high-voltage LiNi<sub>0.5</sub>Mn<sub>1.2</sub>Ti<sub>0.3</sub>O<sub>4</sub> (LNMTO) cathode

- Improve cycle and calendar life by forming a solid-electrolyte interface that effectively passivates the cathode surface
- ▶ Create novel LNMTO core/shell microstructures where Ti is preferentially located at surface and partner with optimized binder/electrolyte chemistries

### ACKNOWLEDGEMENTS

Nexceris would like to acknowledge the excellent support of their project partners, Dr. Jung-Hyun Kim at The Ohio State University, and Dr. James Dong at Navitas Systems.

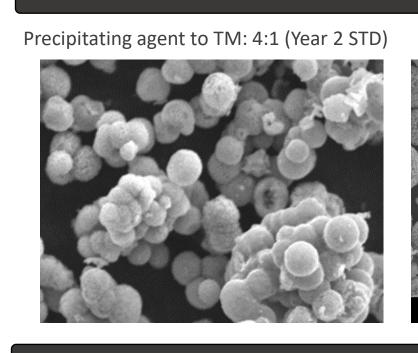
Nexceris would also like to thank the project's DOE Technology Development Manager Dr. Peter Faguy, NETL manager, Adrienne Riggi, and Dr. Jack Deppe and Dr. Ahmad Pesaran for their technical guidance.

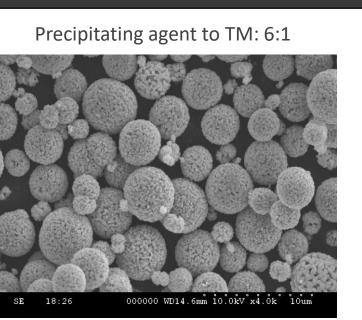
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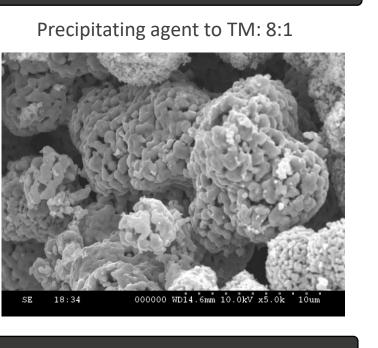
# ACCOMPLISHMENT: Core-Shell Cathode Optimization

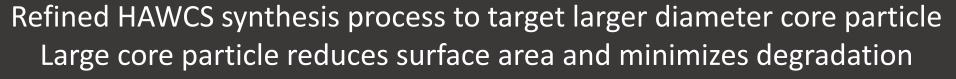
Building on learnings from Year 2 Intermediate 2-Ah cells the HAWCS synthesis process was refined to optimize the LNMO/LNMTO core-shell microstructure

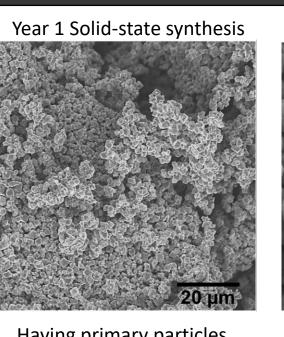
Optimized precipitating agent to transition metal ratio during initial precipitation step Improved LNMO compositional uniformity resulting in higher capacity











(PS analysis of Year 3 powders

Ar-sputtering was performed

Deeper Ti penetrations with

• LNMTO-1: LiNi<sub>0.5</sub>Mn<sub>1.35</sub>Ti<sub>0.15</sub>O<sub>4</sub>

Formation of Ti-rich LiNi<sub>0.5</sub>Ti<sub>1.5</sub>O<sub>4</sub> was found after cycling

LNMTO-3, indicating sacrificial Mn dissolution and

formation of the more-stable Ti-enriched CEI layer

• LNMTO-2: LiNi<sub>0.5</sub>Mn<sub>1.2</sub>Ti<sub>0.3</sub>O<sub>4</sub>

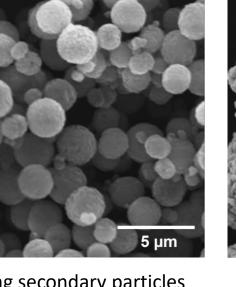
• LNMTO-3: LiNi<sub>0.5</sub>Mn<sub>1.0</sub>Ti<sub>0.5</sub>O<sub>4</sub>

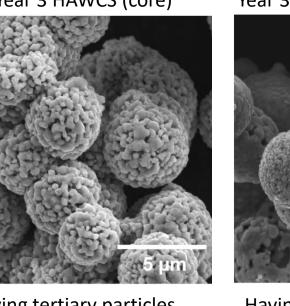
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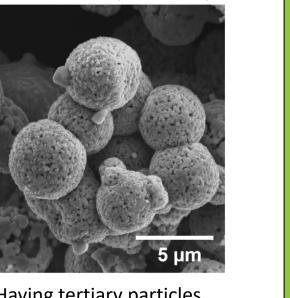
ncreasing Ti in the shell

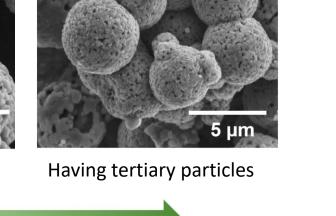
• Bare LNMO: Ti-free

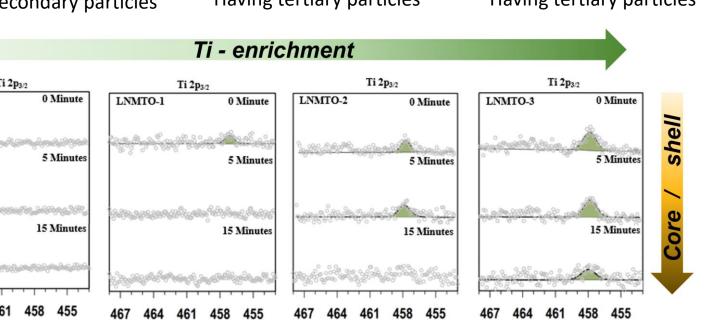
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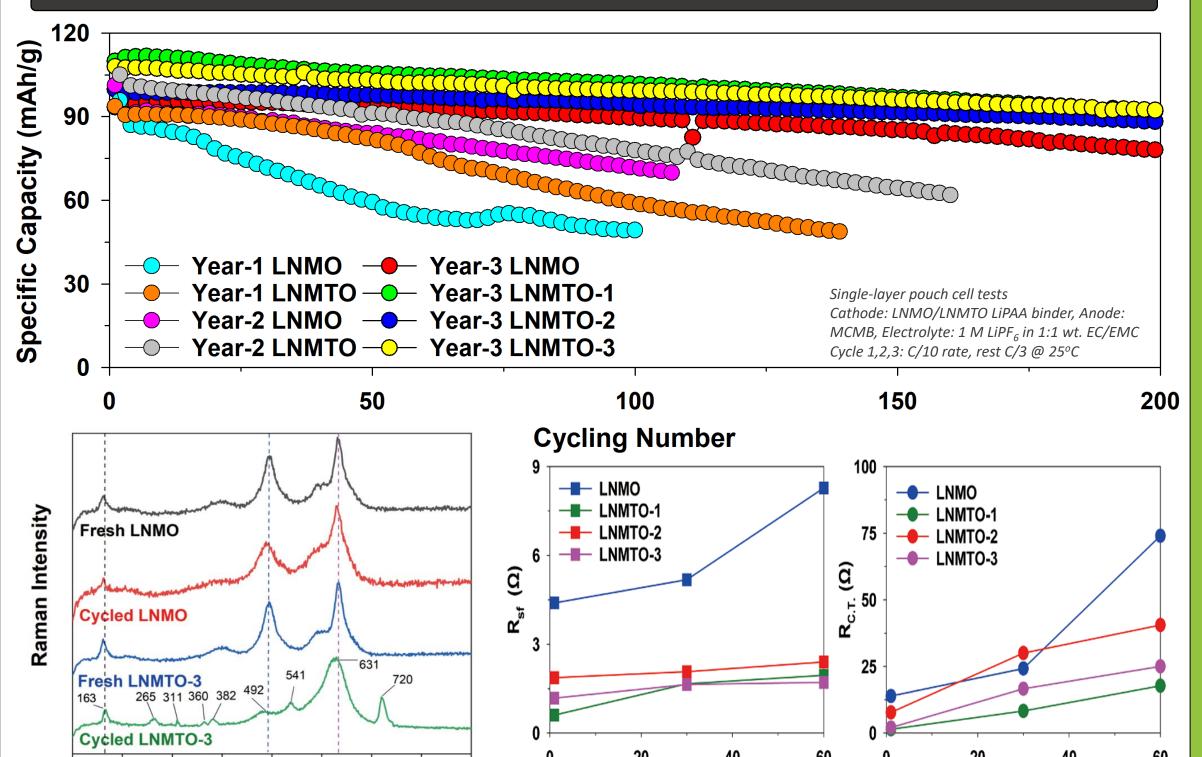




Reduced interfacial impedance ( $R_{cf}$ ) and charge-transfer ( $R_{cT}$ )

resistance with Ti-enriched core-shell cathodes in single-layer pouch

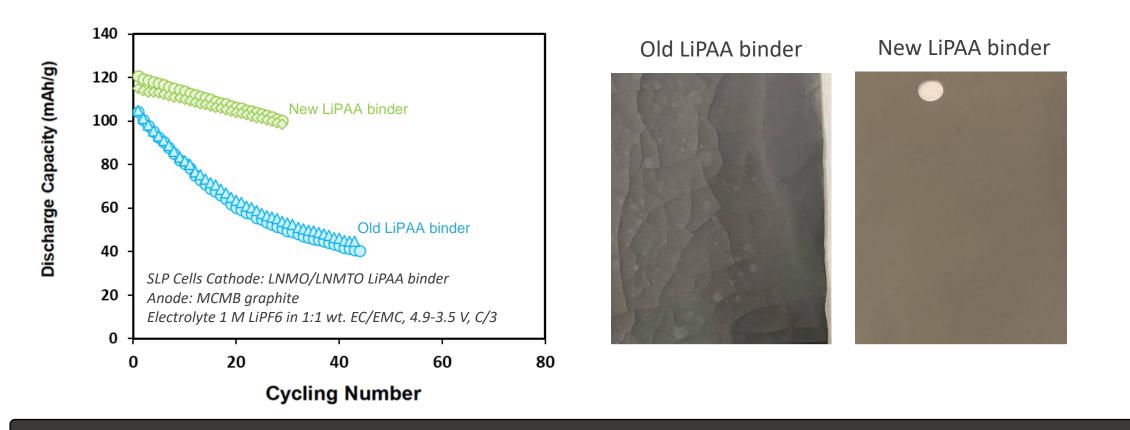
### HAWCS development provided a systematic improvement in performance for both LNMO core and LNMO/LNMTO core-shell cathodes during the project



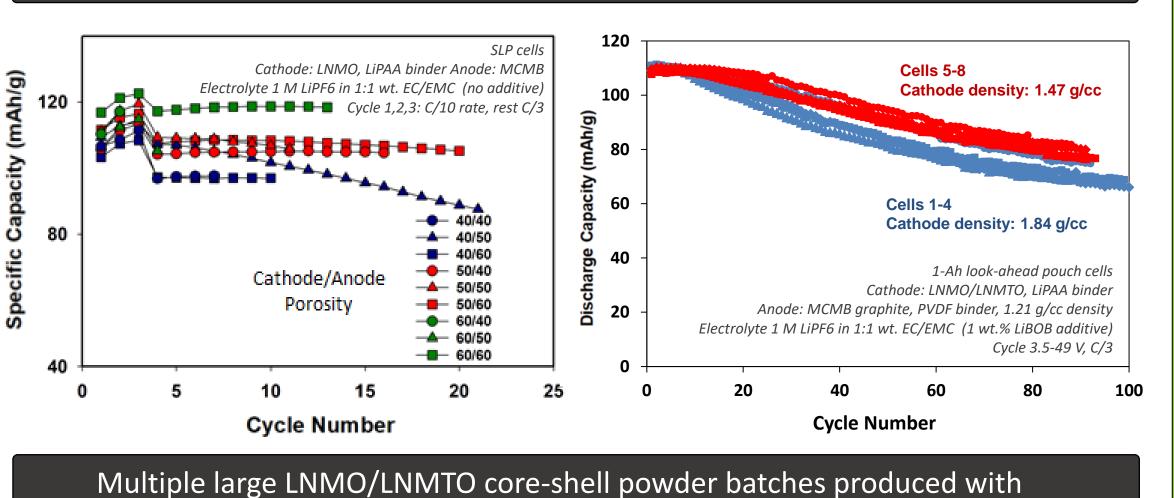
# ACCOMPLISHMENT: 2 Ah PCC Fabrication and Testing

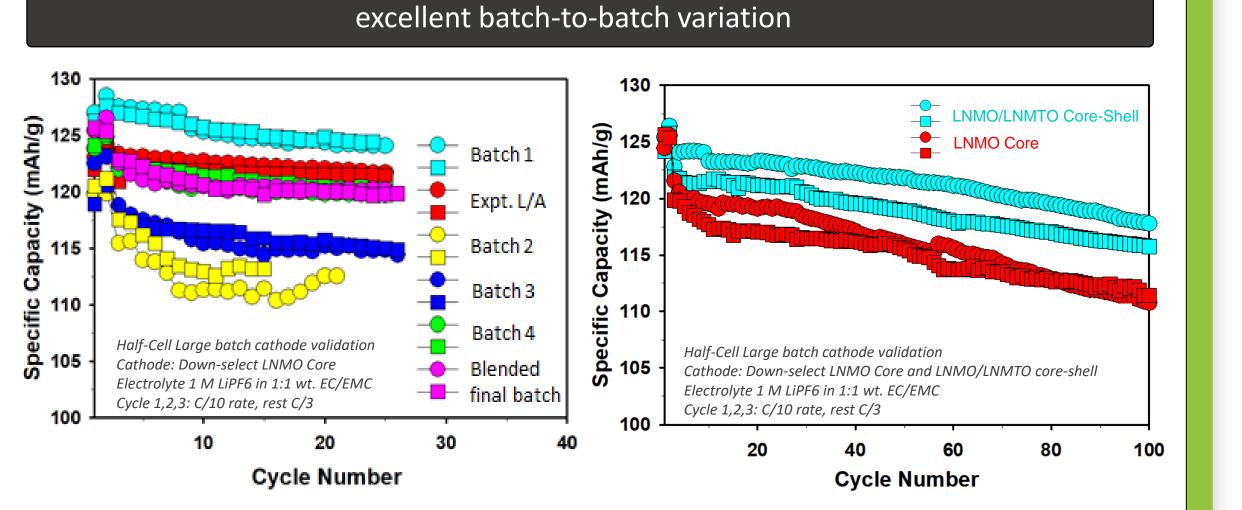
Throughout Year 3 focused on reducing risk through iterative process development to build confidence in scaled powder production larger cell assembly

Resolved discrepancy in SLP performance – segmented to LiPAA binder shelf life Old LiPAA binder makes electrode more susceptible to surface cracks

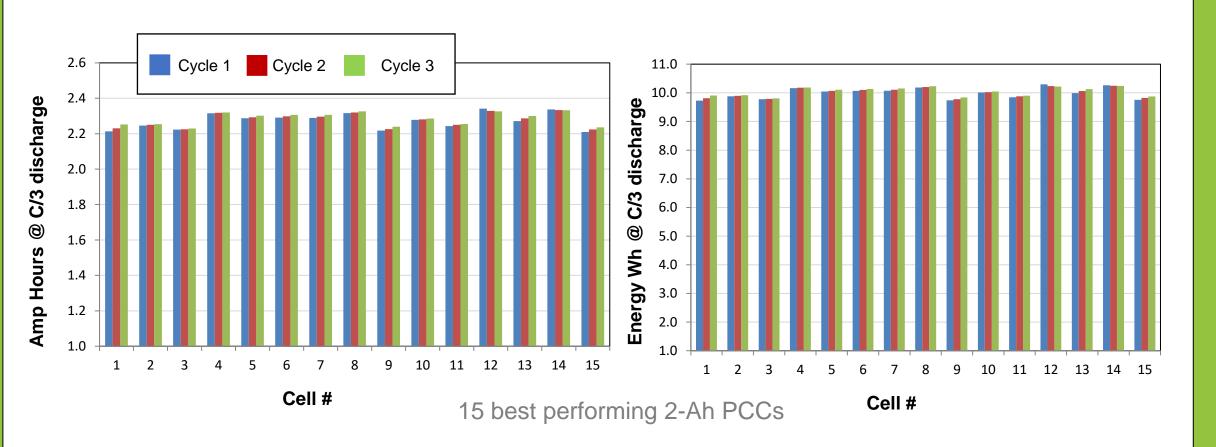


Effect of electrode architecture (density, thickness) studied with SLP cells and lower density (higher porosity) electrodes down-selected





Thirty-four 2-Ah Project Completion Cells (PCCs) successfully fabricated; initial C/3 rated capacity characterized and cells delivered to INL, Navitas and OSU



#### MILESTONE REVIEW

Milestone	Metric	Status
Down-select PCC LNMO/LNMTO core shell cathode powder	1. Multiple process refinements completed to optimize the LNMO/LNMTO core/shell powder Down- selection based on 2-Ah cell testing	<ul> <li>Completed</li> <li>Three LNMO/LNMTO powder iterations completed and PCC powder down selected</li> <li>Cathode powder shipped to Navitas for PCC fabrication</li> </ul>
PCC cell chemistry down- selected	1. Define complementary cell chemistry	Completed  Outilized SLP cell/1-Ah cell formats to identify address cell issues, down-selected cell chemistry/electrode architecture
PCC fabricated	1. Fabricate 30 PCCs to support delivery of 15 to DOE and internal testing.	Completed  OLNMO/LNMTO cathode electrode fabrication completed  OThirty-four 2-Ah PCCs successfully fabricated
PCC tested	1. Testing of PCCs at INL and Navitas/OSU	<ul> <li>Completed</li> <li>Test plan for PCCs finalized with INL</li> <li>PCCs completed C/3 rated capacity/energy cycles</li> <li>Sub-set of cells cycling at Navitas &amp; OSU</li> </ul>

# COLLABORATIONS/COORDINATION OTHER INSTITUTIONS

Nexceris is fortunate to have excellent project partners that have supported cell chemistry development, testing and large cell (2-Ah) manufacture and testing.

Nexceris has continued to work to engage with stakeholders throughout the Li-ion EV battery value-chain to identify commercial opportunities for developed technology

Project Team Member	Relationship
THE OHIO STATE UNIVERSITY	Coin-cell and SLP cell screening of cell chemistries Cell chemistry (additives/binder) development Analytical characterization of cathode materials and electrodes
SYSTEMS	Electrode scale-up, formation-cycling optimization Large format 2-Ahr battery fabrication and testing

## PROPOSED FUTURE RESEARCH

Future work should focus on addressing the challenges identified during this project and meeting the commercial performance and cost targets for the material

Gas generation, leading to cell-swelling was identified as a major degradation mechanism for large-format cells. The existing formation cycling procedures were insufficient to prevent gas generation.

- Optimization of formation cycling to minimize gas generation
- Detection of electrolyte leakage from swollen/repaired cells

Further optimization of the complementary electrolyte formulation (additives) for high voltage cathode chemistries is also recommended.

#### **SUMMARY**

- 1. In budget period 3 the project team refined the novel Hybrid Alternative Wet-Chemical Synthesis (HAWCS) process for reproducibly producing high-quality LNMO high-voltage cathode powder
- 2. Three LNMO/LNMTO core-shell powder iterations were completed, and largebatches of powder delivered to Navitas to address scale-up challenges
- 3. SLP testing identified LiPAA binder shelf-life and electrode architecture (porosity) as critical allowing these parameters to be optimized
- LNMO/LNMTO cathode powder and cell chemistry down-selected and thirty-four 2-Ah cells fabricated